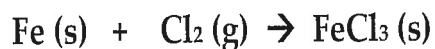
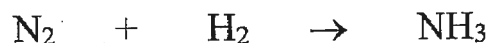


LIMITING REACTANT PRACTICE

1. If 20.00g of iron and 10.0 g of chlorine gas react which reactant is limiting? (balance equation 1st)



2. If 20.00 g of hydrogen gas react with 15.00 g of nitrogen, which gas is limiting? How many grams of ammonia will be produced?



3. 8.34 g of Mg metal reacts with 4.22 g of oxygen gas to produce magnesium oxide. Which reactant is limiting? How many grams of excess reactant will be left over?

